

# **Non-monotonic Relaxation Kinetics of Confined Systems**



**Yaroslav Ryabov**

University of Maryland

**Alexander Puzenko**

**Yuri Feldman**

Hebrew University of Jerusalem

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# Kinetics Models

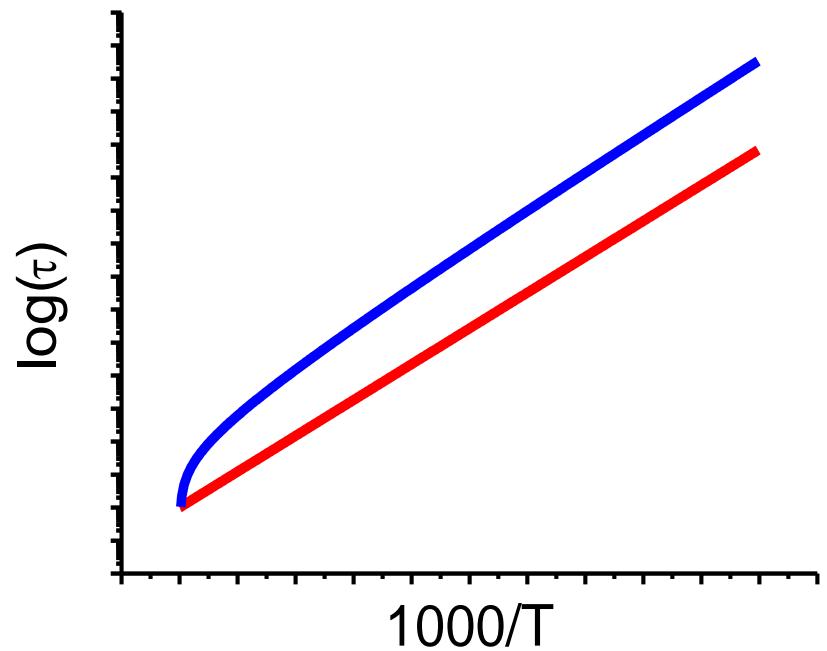
## Transition state theory

**Arrhenius Law (1889)**

$$\tau = \tau_0 \exp\left(\frac{E_a}{kT}\right)$$

**Eyring Law (1935)**

$$\tau = \frac{h}{kT} \exp\left(\frac{\Delta H}{kT} - \frac{\Delta S}{k}\right)$$



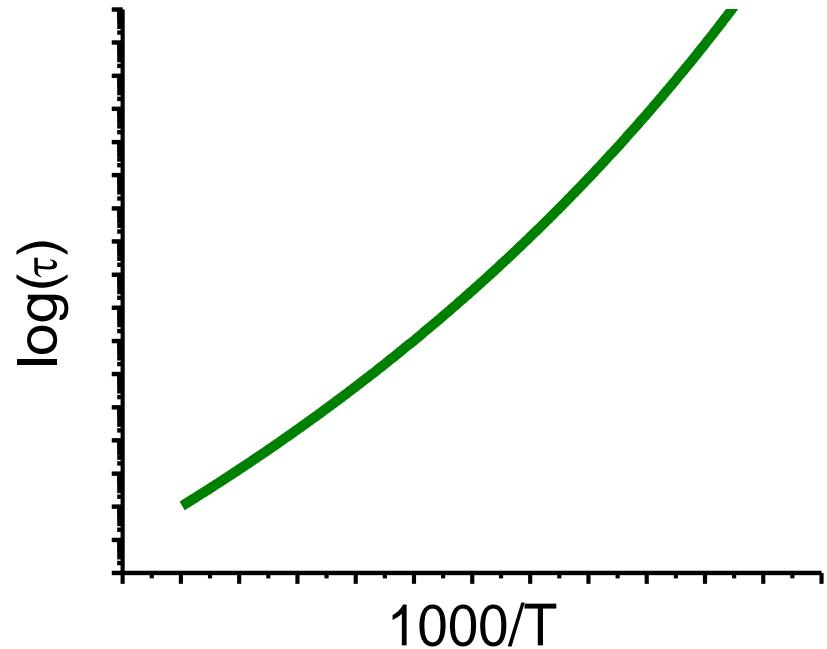
# Kinetics Models

## Free Volume concept

Vogel-Fulcher-Tamman  
VFT (1921,1925,1926)

$$\tau = \tau_0 \exp\left(\frac{DT_k}{T - T_k}\right)$$

$$\tau = \tau_0 \exp\left(\frac{V_0}{V_F}\right)$$
$$V_F \sim T - T_k$$



Fox and Flory (1950)  
Adam and Gibbs (1965)

# Kinetics Models

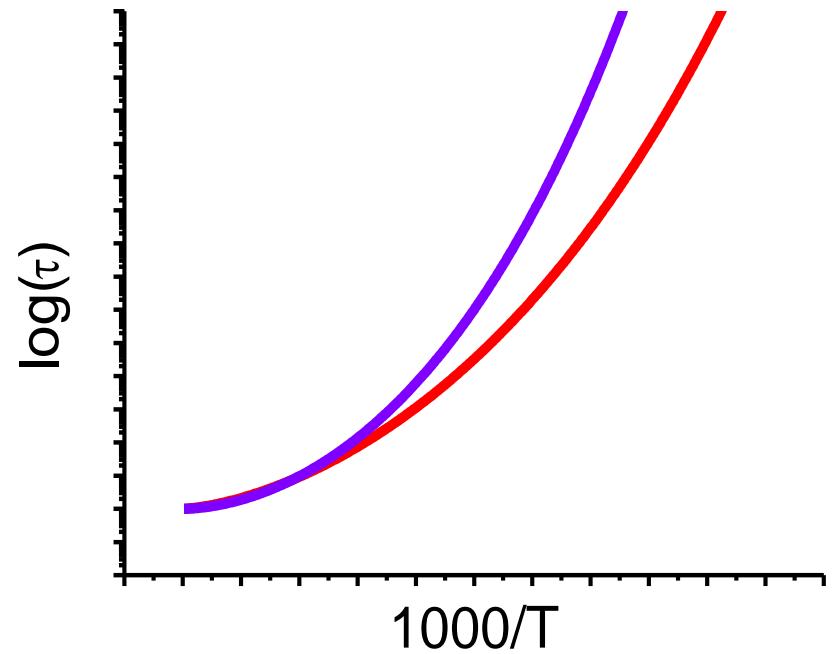
**Bendler and Shlesinger (1987)**  
**Mobile defects in amorphous state**

$$\tau = \tau_0 \exp\left(\frac{B}{(T - T_K)^{3/2}}\right)$$

**Bengtzelius,  
Götze and Sjölander (1984)  
MCT**

$$\tau = \tau_0 \exp\left(\frac{F}{(T - T_K)^\gamma}\right)$$

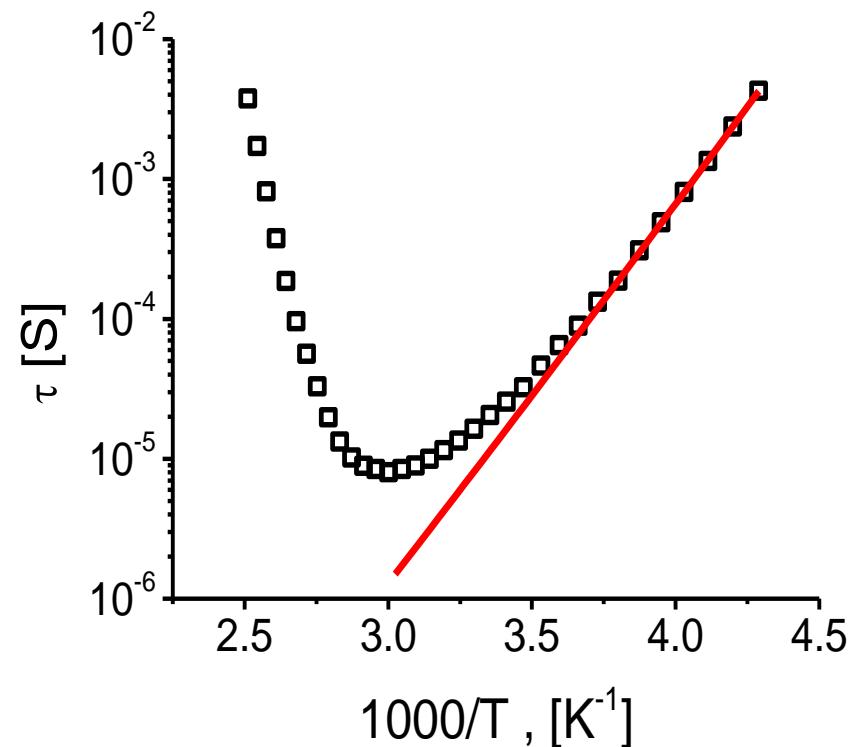
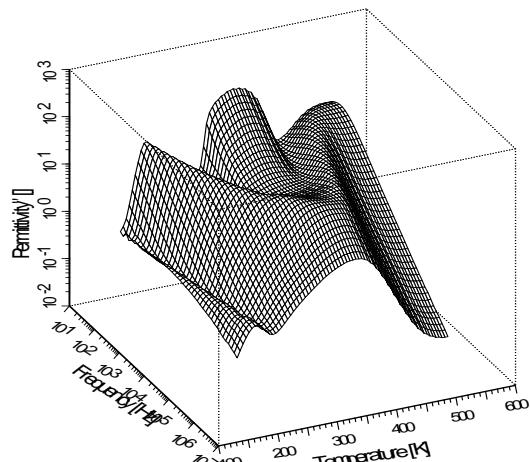
$$\gamma \approx 1.76$$



# Non-monotonic Kinetics

## Experimental observations

**Water confined  
in a porous glass sample**  
**Gutina et al. (1998)**



# Non-monotonic Kinetics

## The Model

$$\frac{1}{\tau} \sim p_1 p_2$$

$$p_1 = \exp\left(-\frac{E_a}{kT}\right) \quad p_2 = \exp\left(-\frac{V_0}{V_F}\right)$$

**Number of active particles**

$$n = n_0 \exp\left(-\frac{E_b}{kT}\right)$$

**Free Volume per particle**

$$\frac{1}{V_F} \sim n_0 \exp\left(-\frac{E_b}{kT}\right)$$

**Total volume of the system**

$$V = \text{const}$$

$$\tau = \tau_0 \exp\left\{\frac{E_a}{kT} + \frac{V_0 n_0}{V} \exp\left(-\frac{E_b}{kT}\right)\right\}$$

# Non-monotonic Kinetics

## The Model

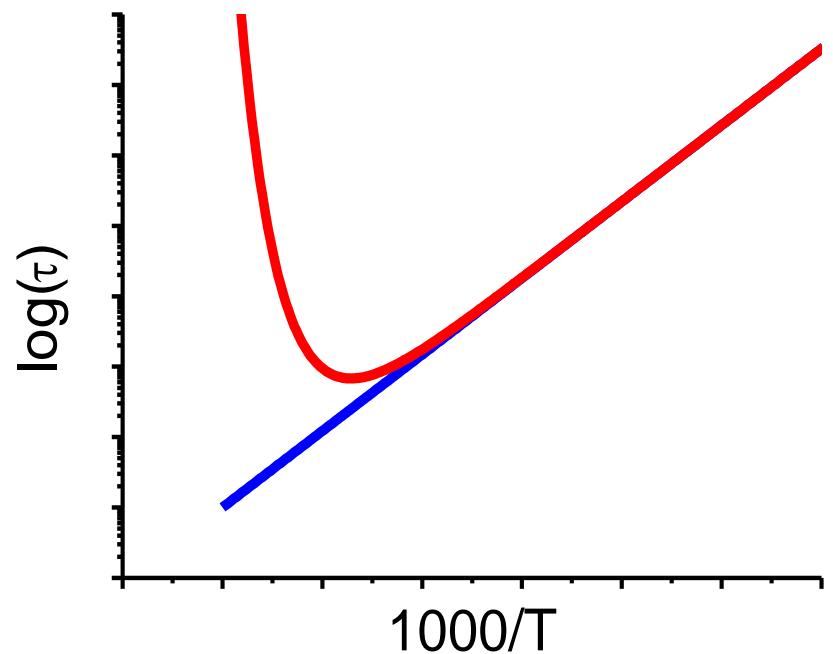
$$\tau = \tau_0 \exp \left\{ \frac{E_a}{kT} + \frac{V_0 n_0}{V} \exp \left( -\frac{E_b}{kT} \right) \right\}$$

$$n_0 \ll \frac{V}{V_0}$$

**Unconfined system  
with Arrhenius kinetics**

$$n_0 \gg \frac{V}{V_0}$$

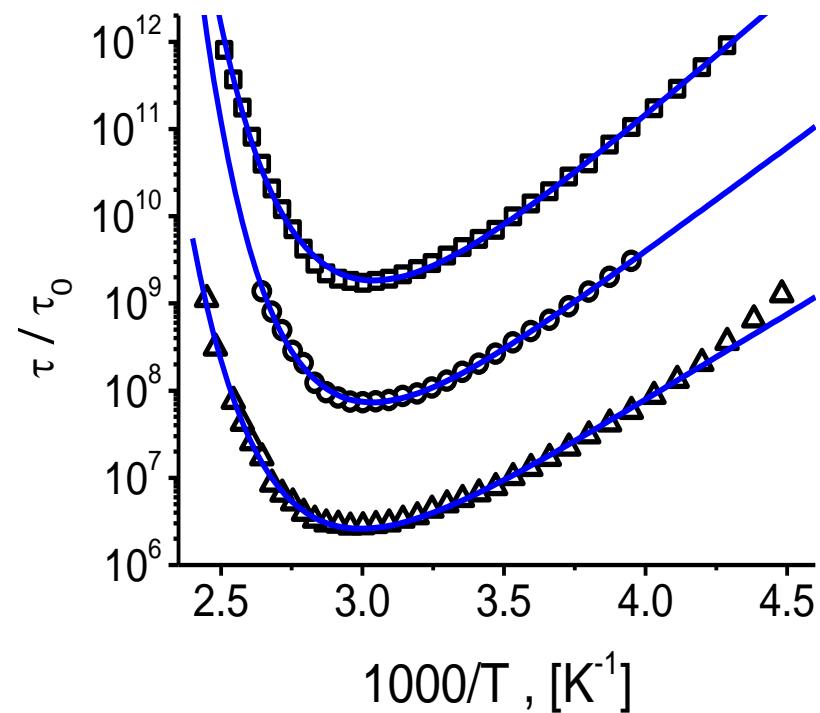
**Confined system  
with non-monotonic kinetics**



# Model VS Experiment

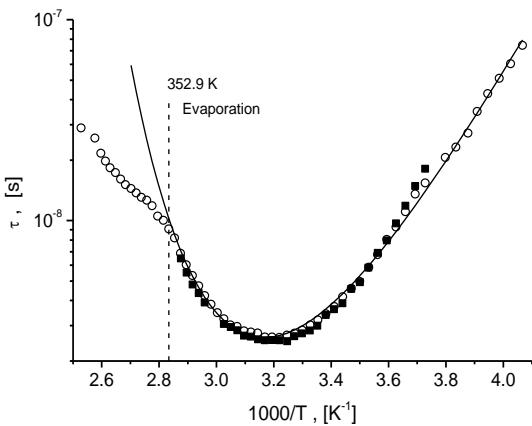
## Water confined in Porous Glasses

Sample	$E_a$ [kJ/mol]	$E_b$ [kJ/mol]	$n_0 V_0 / V$	$\ln \tau_0$
A ○	46	33	$27 \times 10^4$	-27
B □	53	29	$7 \times 10^4$	-33
C △	38	32	$12 \times 10^4$	-26



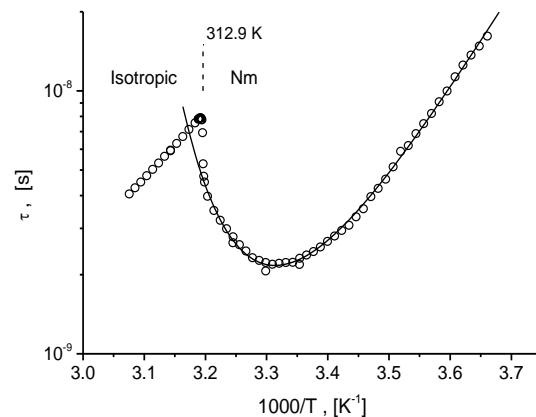
# Model VS Experiment

## Water in Zeolites



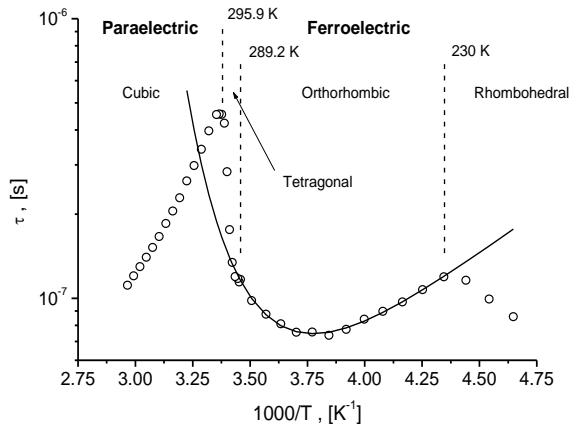
Schönhals et al. (2002)

## Confined 8CB liquid crystal



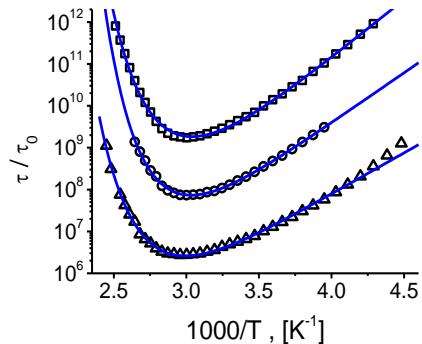
Aliev et al. (2002)

## KTN ferroelectric crystal



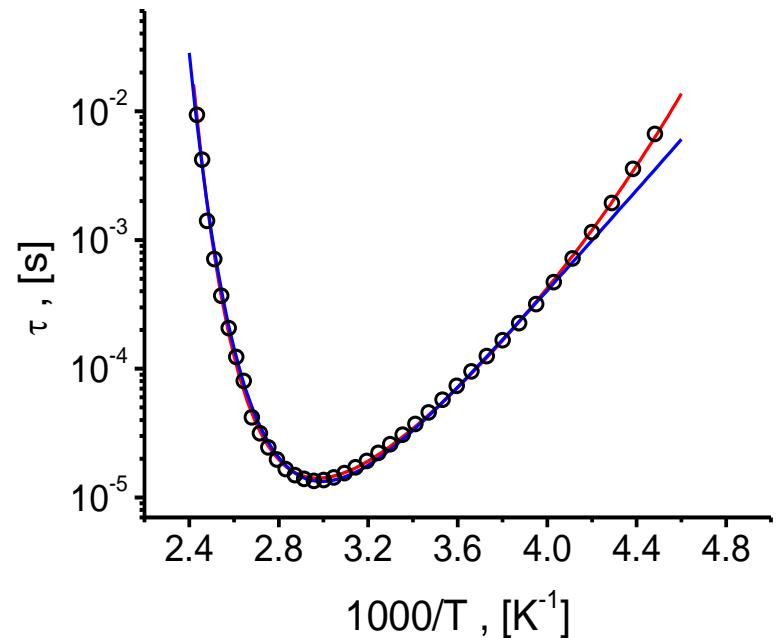
Feldman et al. (2004, 2005)

# Confined Supercooled Water



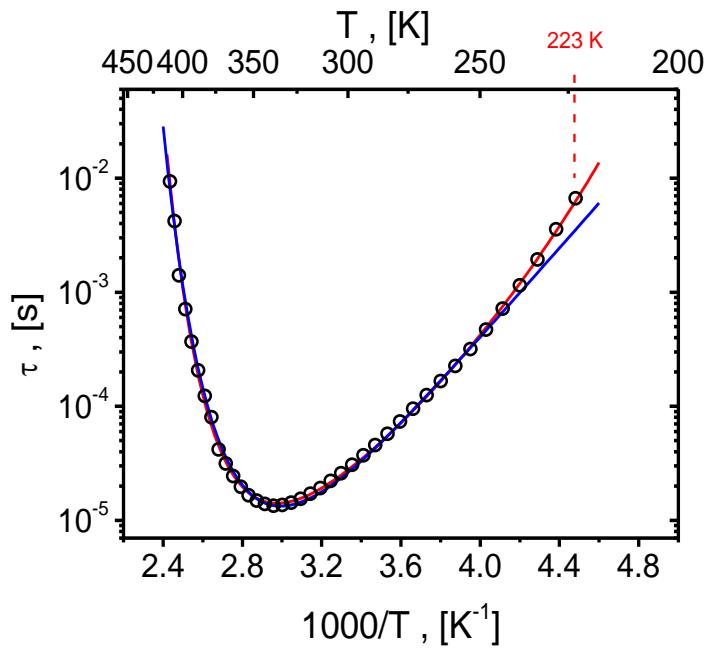
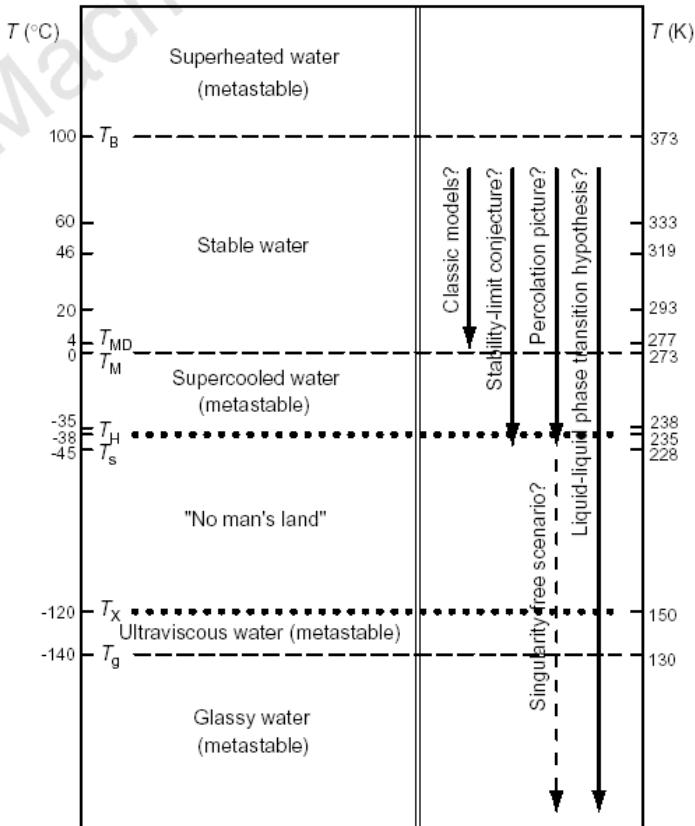
$$p_1 = \exp\left(-\frac{DT_k}{T - T_k}\right)$$

$$\tau = \tau_0 \exp\left\{\frac{DT_k}{T - T_k} + \frac{V_0 n_0}{V} \exp\left(-\frac{E_b}{kT}\right)\right\}$$



# Confined Supercooled Water

Mishima and Stanley (1998)



$$T_k = 124 \pm 7 \text{ [K]} \quad D = 10 \pm 2$$

$$T_g \sim 140 \text{ [K]} \quad D \sim 8$$

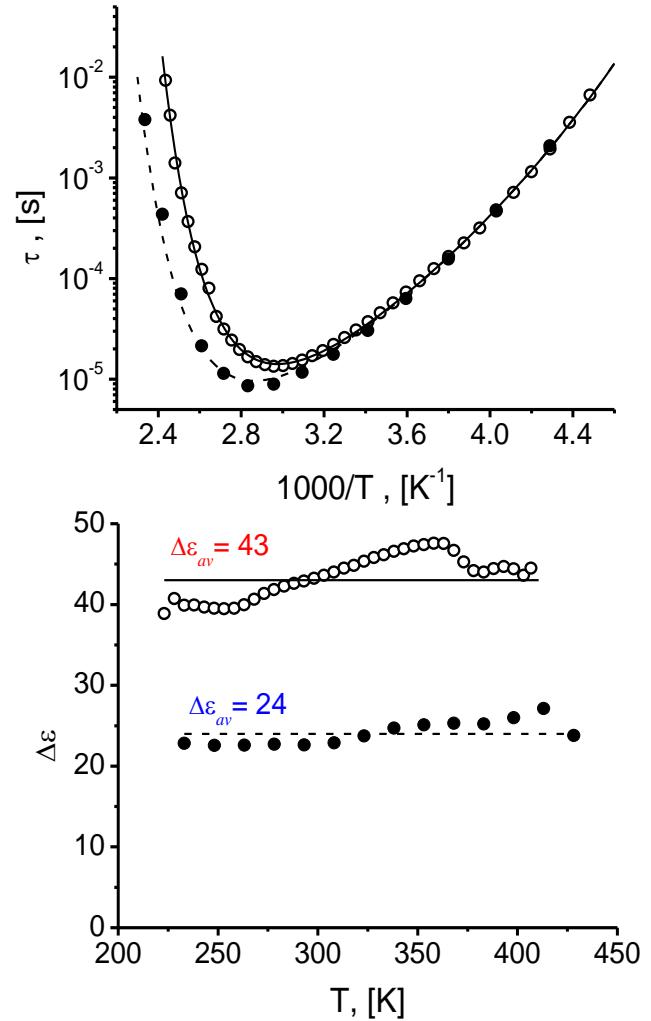
Johary et al. (1987),  
Smith and Kay (1999)

# Confinement Factor

$$\tau = \tau_0 \exp \left\{ \frac{D T_k}{T - T_k} + \frac{V_0 n_0}{V} \exp \left( -\frac{E_b}{kT} \right) \right\}$$

$$\frac{V_0 n_0}{V} \quad n_0 \sim \Delta \varepsilon$$

$$\frac{\Delta \varepsilon_{av}}{\Delta \varepsilon_{av}} = 1.8$$



# Protein folding kinetics

## Random Energy Model

Derrida (1980)

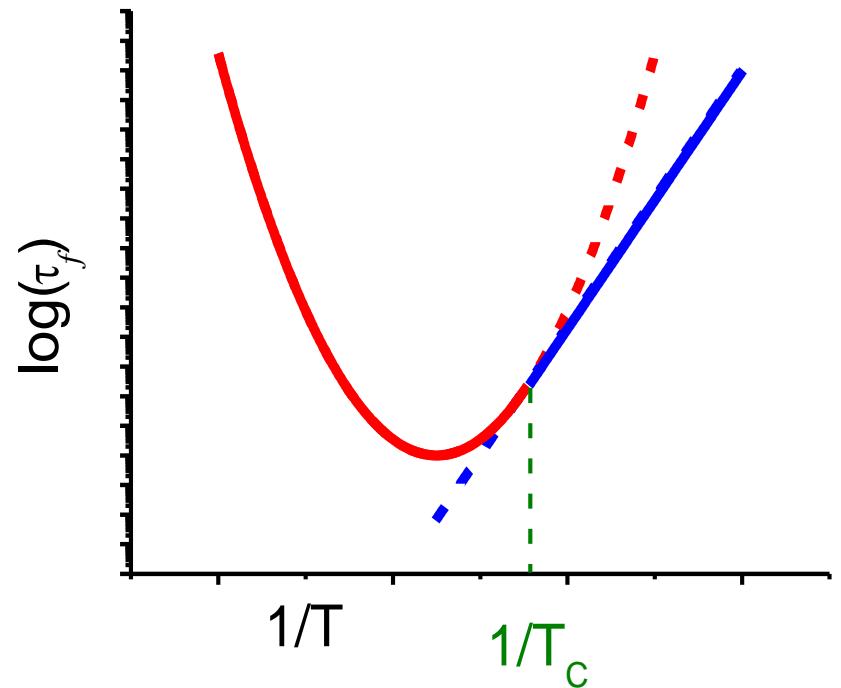
$$n(E) = \frac{\Omega}{\sqrt{2\pi\Sigma^2}} \exp\left(-\frac{(E - \bar{E})^2}{2\Sigma}\right)$$

Bryngelson and Wolynes (1987, 1989)

$$\ln\left(\frac{\tau_f}{\tau_0}\right) = \frac{E^*}{kT} + \frac{E_c T_c}{2k^2} \left( \frac{1}{T} - \frac{1}{T_c} \right)^2 \quad T > T_c$$

$$\ln\left(\frac{\tau_f}{\tau_0}\right) = \frac{E^*}{kT} \quad T < T_c$$

$$E_c = \bar{E} - \Sigma \sqrt{1 \ln \Omega}$$



# Protein folding kinetics

## Confinement in Conformation space

$$p_f = p_b^q p_n^{qN}$$

Probability to have  
an ‘active’ bond

$$p_b = \exp\left(-\frac{E_b}{kT}\right)$$

Probability to find  
an ‘proper’ connection

$$p_n = \exp\left(-\frac{\omega_0}{\omega_f}\right)$$

Total conformation volume

$$\Omega_0 = q^N$$

$$\tau_f \sim \frac{1}{p_f} \sim q \frac{E_b}{kT} + qN \exp\left(-\frac{E_b}{kT}\right)$$

Free conformation volume

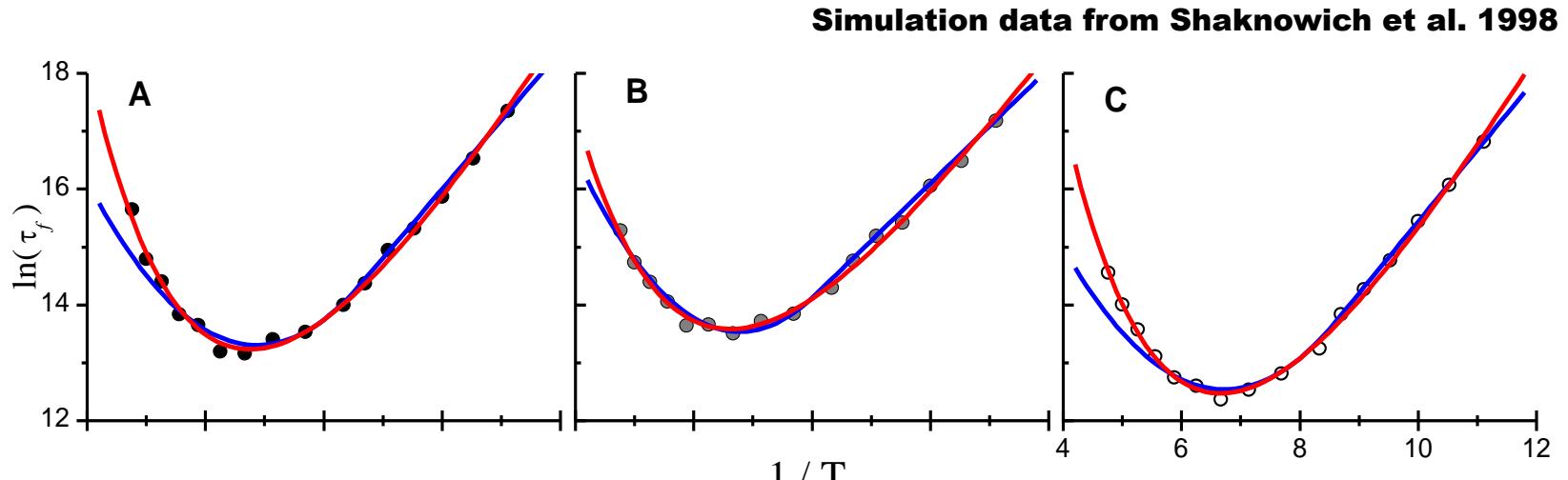
$$\omega_f = \frac{\Omega_0}{N_d}$$

$$\omega_0 = \frac{\Omega_0}{qN}$$

$$\omega_f = qN \exp\left(\frac{E_b}{kT}\right)$$

# Protein folding kinetics

## Models VS simulation data



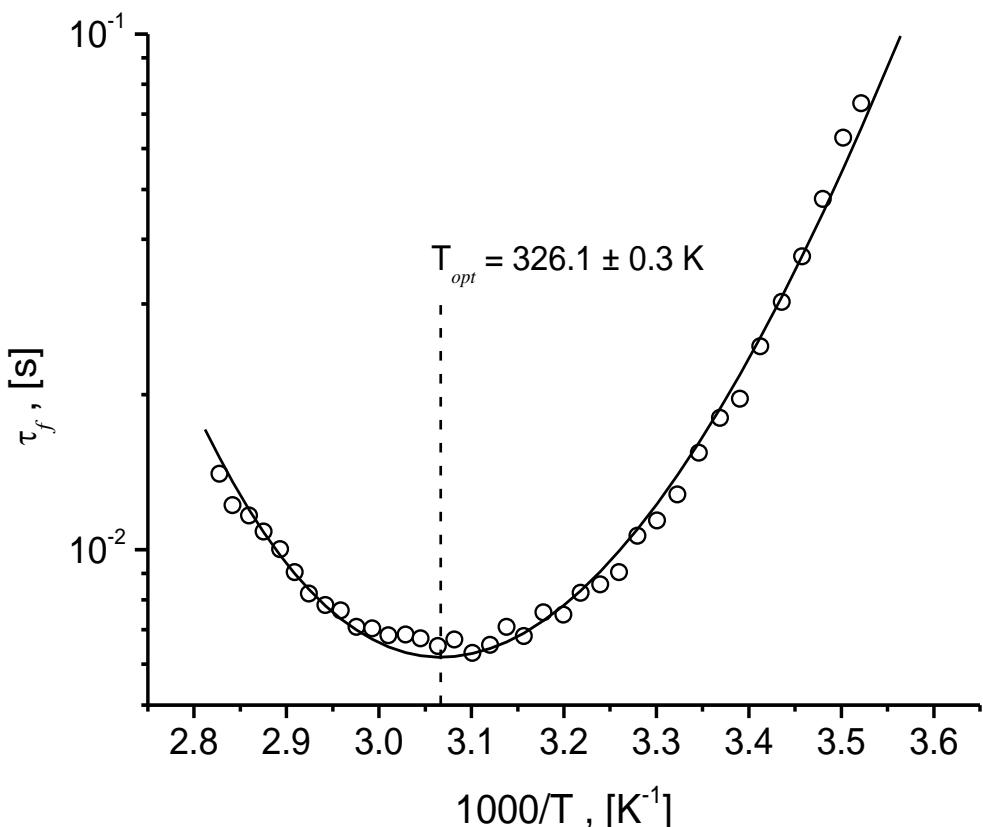
$N = 27$

	REM					Confinement Model			
	$\ln \tau_0$	$E_c$	$E^*$	$1/T_C$	$\chi^2$	$\ln \tau_0$	$E_b$	$q$	$\chi^2$
A	4.20	-5.92	1.19	8.60	0.077	-1.18	0.49	3.26	0.010
B	6.10	-6.32	0.99	58.04	0.016	1.84	0.50	2.73	0.008
C	3.04	-5.68	1.24	8.60	0.083	2.14	0.49	3.40	0.006

# Protein folding kinetics

## Model VS experimental data

Data from Fersht et al. 1996 (CI2)



$$\ln \tau_0 = -83$$

$$E_b = 11.3$$

$$qN = 967$$

**More experimental data needed**

# Protein folding kinetics

## Levintahal's Paradox (1969)

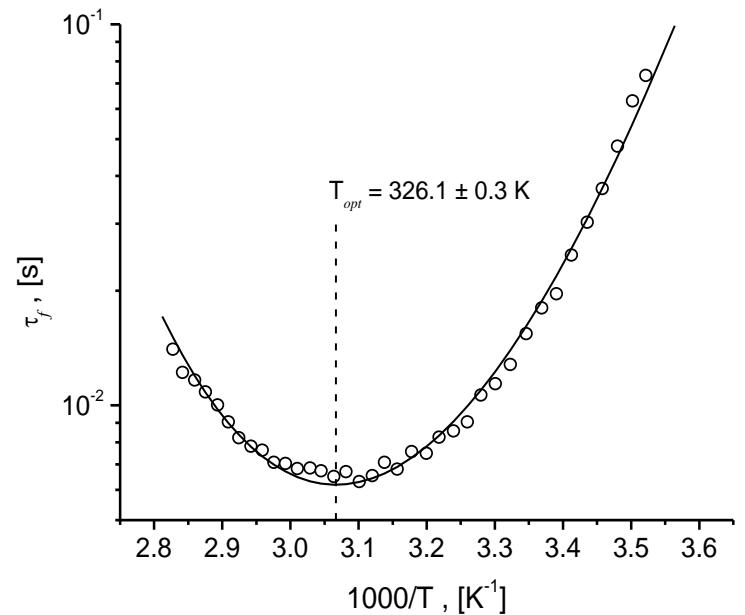
$$\tau_f = \tau_0 q^N$$

**Algebraic scaling**

$$\tau_{fast} = \tau_0 (eN)^q$$

$$q \sim 2 \div 3$$

**Camacho and Thirumalai (1993)**



**Have To be tested**

$$T_{opt} = \frac{E_b}{k \ln N}$$

# Conclusions



- **Non monotonous kinetics can be a result of confinement either in real or in conformational space**
- ✓ **Confined water can be supercooled below its homogeneous nucleation point which opens new possibilities for investigation of water glass state**
- ✓ **The concept of confinement in conformational space can serve as a paradigm for protein folding kinetics, which leads to known algebraic scaling for protein folding times.**

# Thanks



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